

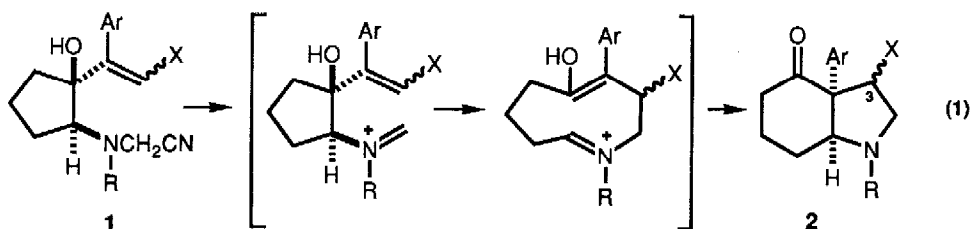
PREPARATION OF FUNCTIONALIZED HYDROINDOL-3-OLS VIA TANDEM AZA-COPE
REARRANGEMENT-MANNICH CYCLIZATIONS. FORMAL TOTAL SYNTHESIS OF
(+)-6a-EPIPRETAZETTINE AND RELATED ALKALOIDS

Larry E. Overman* and Hanno Wild¹

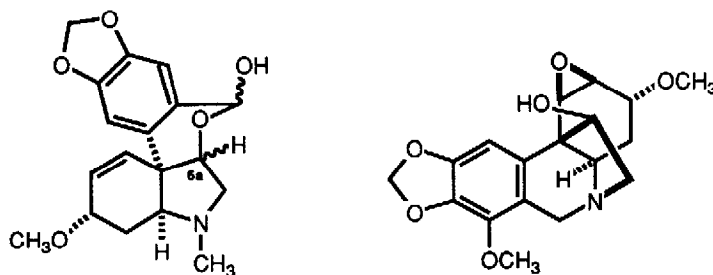
Department of Chemistry, University of California
Irvine, California 92717

Summary: Key steps in this approach to *amaryllidaceae* alkaloids containing oxidation at the β -position of the pyrroline ring are (a) copper(II)-promoted rearrangement of aminocyclopentanol **11** to afford the 3-silylhydroindole **12**, and (b) oxidation of **12** by the general procedures of Kumada and Fleming to give the *cis*-3a-aryl-3-hydroxyhexahydroindolone **13** in 71% overall yield from **11**.

We have reported an efficient method for preparing *cis*-3a-arylhydroindoles **2** that involves a tandem aza-Cope rearrangement-Mannich cyclization reaction (eq 1) as its key element.² This chemistry



constitutes a general approach for alkaloid synthesis, since the *cis*-3a-arylhydroindole ring system is found in a number of alkaloids, and has been utilized by us to achieve total syntheses of *amaryllidaceae*,³ *aspidosperma*,⁴ and *melodinus*⁵ alkaloids. A number of alkaloids, including the most pharmacologically active members of the crinine class of *amaryllidaceae* alkaloids (e.g. **3** and **4**),⁶ contain *cis*-3a-arylhydroindole rings that are oxidized at the β -carbon of the pyrrolidine ring.⁷ Not surprisingly, oxygen functionality of this type is not compatible with the cationic aza-Cope rearrangement-Mannich cyclization reaction (eq 1, X=OH or OR).⁸ We have investigated, therefore, the transformation illustrated in eq 1

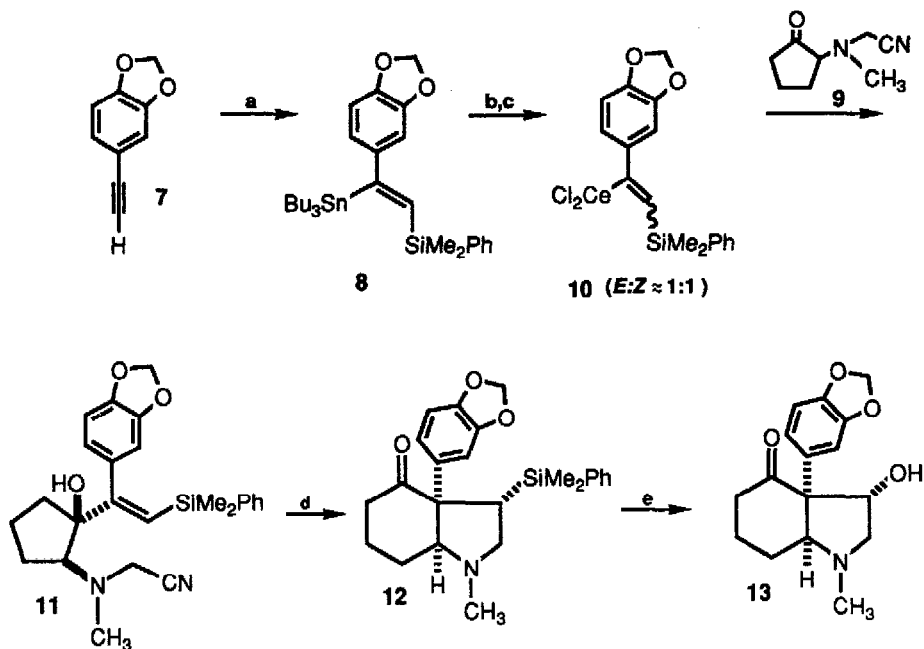


3 α -6a-H, pretazettine

5 β -6a-H, 6a-epipretazettine

4, cavinine

SCHEME I

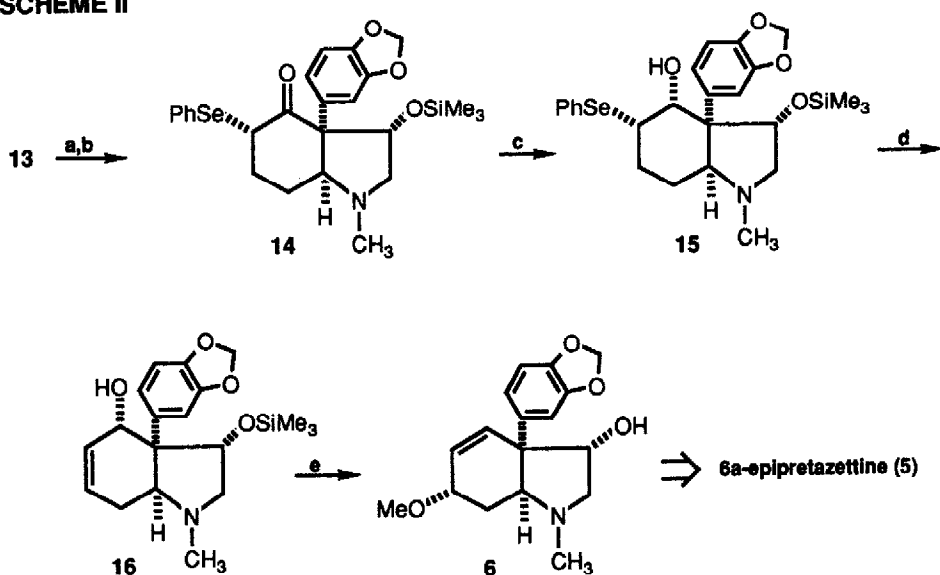


(a) $\text{Bu}_3\text{SnSiMe}_2\text{Ph}$, $\text{Pd}(\text{Ph}_3\text{P})_4$, THF, 60°C , 8h (95%); (b) $n\text{-BuLi}$, THF, -78°C , 1h; (c) 0°C , 5 min, CeCl_3 ; **9**, -78°C (25% from **8**);
 (d) $\text{Cu}(\text{OTf})_2$, THF, reflux, 2h (94%); (e) $\text{HBF}_4\cdot\text{OEt}_2$, CH_2Cl_2 , rt, 2h; H_2O_2 , KF, DMF, 60°C , 7h (78%).

where X is a silyl substituent, since Kumada⁹ and Fleming¹⁰ have developed versatile sequences for the conversion of $\text{C-SiR}_3 \rightarrow \text{C-OH}$. In this communication we report that the "hydroxyl equivalency" of the PhMe_2Si group¹⁰ can be exploited to prepare oxidized *cis*-3 α -arylhydroindoles **2** ($\text{X}=\alpha\text{-OH}$) by the aza-Cope rearrangement-Mannich cyclization strategem. Specifically, we describe the synthesis of **6** (see Scheme II), a late intermediate in incisive total syntheses of (+)-6 α -epipretazettine and (+)-tazettine achieved by the Danishefsky group.¹¹

Our investigations began with silyl stannane **8**^{12a} which was prepared from (3,4-methylenedioxy)phenylacetylene (**7**)¹³ and (dimethylphenylsilyl)tributylstannane by the general method of Chenard and Van Zyl.¹⁴ We initially examined the reaction of readily available aminocyclopentanone **9**³ with the vinyl lithium¹⁵ and vinyl cerium¹⁶ reagents derived from **8** in the hope of preparing a cyclization substrate with the *trans*-orientation of aryl and silicon functionality. Unfortunately, enolization of **9** predominated under all conditions examined, presumably reflecting the steric bulk of these organometallic intermediates which have the silyl group *cis* to the metal and the well-known susceptibility of cyclopentanones to deprotonation.¹⁷ On the other hand, the (*E*)-cerium reagent, which could be formed from **8** as a 1:1 mixture of stereoisomers by equilibration of the corresponding lithium intermediate at 0°C , reacted with **9** at -78°C in modest yield (41% based on consumed **9**, 25% actual yield) to provide a single adduct **11**^{12a} whose stereochemistry was ascertained by ¹H NMR NOE experiments.

SCHEME II



(a) Me_3SiOTf , NEt_3 , CH_2Cl_2 , 0°C , 1h (95%); (b) PhSeCl , CH_2Cl_2 , 0°C , 1h (78%); (c) $i\text{-Bu}_2\text{AlH}$, toluene, -78°C , 15 min (81%); (d) $m\text{-ClC}_6\text{H}_4\text{CO}_3\text{H}$, CH_2Cl_2 , -78°C , 45 min; xylene, reflux, 15 min (81%); (e) $(\text{CH}_3\text{SO}_2)_2\text{O}$, NEt_3 , THF, 0°C , 20 min; MeOH , NEt_3 , reflux, 15 min; $n\text{-Bu}_4\text{NF}$, THF, rt, 15 min (25%).

The key rearrangement of 11 was accomplished by treatment with 2 equiv of copper(II)triflate in refluxing tetrahydrofuran (THF) to give the silyl hydroindolone 12^{12a,b}, mp 129°C , in 94% yield. Conversion of 12 to the fluorosilane,¹⁰ followed by oxidation of this intermediate at 60°C with 20 equiv of 30% H_2O_2 in the presence of 5 equiv of KF^9 gave cleanly the desired hydroxylated hydroindolone 13^{12a,c} in 76% yield. Significantly, the basic nitrogen did not require protection during this oxidative transformation.

The conversion of 13 to the Danishefsky intermediate 6¹¹ is summarized in Scheme II. Of note in this sequence was the reduction of the selenoketone 14 with $i\text{-Bu}_2\text{AlH}$ which provided exclusively the α alcohol 15.^{12a,18} The steric bulk of the angular aryl and α -phenylseleno groups is apparently sufficient to direct the reducing agent to the concave face of this *cis*-fused bicyclic ketone. The conversion of 16^{12a} to 6 was accomplished by methanolysis of the allylic mesylate intermediate following a general strategy defined earlier by Whitlock¹⁹ and Danishefsky.¹¹ Hydroindole 6 prepared in this fashion was indistinguishable (^1H NMR, IR, mass spec, and TLC comparisons in three solvents) with an authentic sample.

In summary, the efficient (71% overall yield) two-step conversion of aminocyclopentanol 11 to the *cis*-3a-aryl-3-hydroxyhexahydroindolone 13 demonstrates that a silicon substituent can be employed as an alcohol surrogate in the stereocontrolled synthesis of hydroxylated hydroindoles by the tandem aza-Cope rearrangement-Mannich cyclization method. For this sequence to be a practical route to these intermediates, more efficient methods for assembling the silicon-containing rearrangement precursors will be required.

Acknowledgment. The support of this research by the Alexander von Humboldt Foundation and PHS Grant NS-12389 is gratefully acknowledged. We particularly wish to thank Professor Danishefsky for providing a comparison sample of **6**.

REFERENCES AND NOTES

1. Feodor Lynen Postdoctoral Fellow at the Alexander von Humboldt Foundation, 1987-1988.
2. Jacobsen, E.J.; Levin, J.; Overman, L.E. *J. Am. Chem. Soc.* **1988**, *110*, 4329 and references cited therein.
3. Overman, L.E.; Mendelson, L.T.; Jacobsen, E.J. *J. Am. Chem. Soc.* **1983**, *105*, 6629. Overman, L.E.; Sugai, S. *Helv. Chim. Acta* **1985**, *68*, 745.
4. Overman, L.E.; Sworin, M.; Burk, R. M. *J. Org. Chem.* **1983**, *48*, 2685.
5. Overman, L.E.; Robichaud, A.J.; Robertson, G. to be submitted.
6. Pretazettine (**3**) exhibits a broad biological profile including activity against Rauscher leukemia, while cavinine (1,2- β -epoxyambelline **4**) is an immunostimulant.⁷
7. For a recent authoritative review covering structure, synthesis and biological activity of *amaryllidaceae* alkaloids, see Martin, S.F. *Alkaloids(N.Y.)* **1987**, *30*, 252.
8. Elimination of X predominates, see Okazaki, M. *Ph.D. Thesis*, University of California, Irvine, 1987.
9. Tamaro, K.; Ishida, N.; Tanaka, T.; Kumada, M. *Organometallics* **1983**, *2*, 1694.
10. Fleming, I.; Henning, R.; Plaut, H. *J. Chem. Soc. Chem. Commun.* **1984**, 29.
11. Danishefsky, S.; Morris, T.; Mullen, G.; Gammill, R. *J. Am. Chem. Soc.* **1982**, *104*, 7591.
12. (a) All new compounds exhibited spectroscopic properties (¹H NMR, ¹³C NMR, IR, high resolution MS) in accord with their assigned structures. (b) The molecular composition of this intermediate was confirmed by combustion analysis. (c) Characterization data for **13**: ¹H NMR (300 MHz, CDCl₃) δ 4.79 (dd, J = 6.1, 1.9 Hz, H_{3 β}), 3.52 (dd, J = 10.6, 6.2 Hz, H_{2 β}), 3.29 (broad s, H_{7 a}), 2.28 (s, NCH₃), 2.24 (dd, J = 10.7, 2.0 Hz, H_{2 α}); ¹³C NMR (75 MHz, CDCl₃) 210.3, 148.8, 147.4, 128.8, 121.8, 109.2, 108.8, 101.9, 72.8, 68.0, 65.8, 62.1, 39.7, 39.4, 22.6, 22.0; IR (CCl₄) 3596, 1708 cm⁻¹; MS (EI) m/z 289.1300 (78%, 289.1314 calcd. for C₁₆H₁₉NO₄).
13. Prepared in 78% yield from piperonal by the general method of Corey and Fuchs: Corey, E.J.; Fuchs, P.L. *Tetrahedron Lett.* **1972**, 3769.
14. Chenard, B.L.; Van Zyl, C.M. *J. Org. Chem.* **1986**, *51*, 3561.
15. In contrast to the 2-silyl-1-phenylvinyl lithium reagents described in ref 14, this vinylolithium intermediate was configurationally stable at -78°C for 2.5 h.
16. Imamoto, T.; Sugiura, Y.; Takiyama, N. *Tetrahedron Lett.* **1984**, *25*, 4233.
17. The (*E*)-vinylolithium reagent derived from **8** was successfully added to cyclohexanone to give the corresponding tertiary alcohol in 46% yield. Enolization with **9** was confirmed by quenching reaction mixtures with D₂O.
18. The stereostructure of **15** followed from its 500 MHz ¹H NMR spectrum (CDCl₃): δ 4.46 (dd, J = 5.4, 2.5 Hz, H_{3 β}), 3.93 (dd, J = 11.6, 4.1 Hz, H_{4 β}), 3.13 (broad s, H_{5 β}), 2.60 (d, J = 11.7 Hz, OH).
19. Whitlock, H.W.; Smith, G.L. *J. Am. Chem. Soc.* **1967**, *89*, 3600.

(Received in USA 5 October 1988)